

CO₂ geological storage: A review on present and future prospects

Umer Zahid, Youngsub Lim, Jaeheum Jung, and Chonghun Han[†]

School of Chemical and Biological Engineering, Seoul National University, Seoul 151-744, Korea
(Received 31 August 2010 • accepted 12 October 2010)

Abstract—CO₂ can be stored in geological media for hundreds to thousands of years depending on the location and trapping mechanism(s) involved. A saline aquifer presents the largest capacity available for CO₂ storage among all geological storage options. Two main methodologies proposed by the Department of Energy, US (DOE) and carbon sequestration leadership forum (CSLF) are used for capacity estimation of geological locations for CO₂ storage. A study conducted by Global CCS Institute in 2010 identified 80 large scale integrated projects which will prove to be a huge step in building confidence and commercialization of storage projects in the near future. Use of reliable monitoring tools and accurate simulation software is a must for safe and cost-effective CO₂ storage.

Key words: Carbon Dioxide, Geological Storage, Capacity Estimation, Sequestration Projects, Monitoring

INTRODUCTION

Carbon capture and storage (CCS) has advanced towards commercialization, especially through the commissioning of CCS pilot plants, continued learning from projects already in operation and the development of legal and regulatory frameworks. While much progress has been achieved, the recommendation made by the G8 Leaders at the 2008 Hokkaido Toyako Summit that 20 large scale CCS demonstration projects should be launched by 2010 remains a challenge [1]. Storage being the integral part for implementation of CCS is still in a progressing phase. Many researchers and organizations are studying various storage sites and steps needed to assure a safe and cost-effective storage site. Michael et al. [2] presented a review of the existing saline aquifers storage projects with key technical site data, monitoring techniques used and injection strategies. Enhanced oil recovery (EOR) is a mature technique, but interest has increased with the CO₂ storage opportunities. In the United States, approximately 73 EOR operations inject up to 30 Mton of CO₂ per year. CO₂ storage in coal seams is an emerging technique and in demonstration phase [3]. Reeves et al. [4-6] have summarized experience and results from the Allinson enhanced coal bed methane (ECBM) pilot plant, the world's first ECBM project. Many other researchers [7-11] have studied and explained regional CO₂ storage in coal seams. All of these studies are focused on one specific storage option. The purpose of this study is to present an up-to-date overall review of all geological storage options, including a summary of the CO₂ storage mechanism at different locations, previous CO₂ storage projects, capacities calculation methods, monitoring techniques deployed at storage sites and future prospects of storage projects. This study also presents the importance of simulation tools used for modeling the storage sites and comparison of CO₂ storage cost estimates of previous researches.

TRAPPING MECHANISMS

CO₂ is injected usually in the supercritical form into the saline aquifer or depleted oil or gas reservoir. The point at which CO₂ transforms from critical to supercritical point is 31.1 °C and 7.38 MPa [12-16]. Most of the geological formations consist of layers of porous rocks capped by a non-porous rock(s) above them. Once injected, supercritical CO₂ is 30-40% less dense than a typical aqueous saline [17,18]. The supercritical CO₂ tends to be buoyant and will rise upward through the rock until it encounters a barrier of non-porous rock. In supercritical form, a larger volume of CO₂ can be stored in the pore space available [19,20]. CO₂ can be trapped by a number of different mechanisms and depend on specific geological conditions.

- Structural or stratigraphic traps, typical of hydrocarbon accumulations [21]
- Hydrodynamic traps, where the dissolved and immiscible CO₂ travels with the formation water for very long residence (migration) times [22]
- Residual gas trapping, where the CO₂ becomes trapped in the pore spaces by capillary pressure forces [17,23,24]
- Solubility trapping, where the CO₂ dissolves into the formation water [25,26]
- Mineral trapping, where the CO₂ precipitates as new carbonate minerals [8,27]
- Adsorption trapping, where the CO₂ adsorbs onto the surface of coal [3,8]

Time is also an important factor in the long-term storage and behavior of CO₂. As the time goes on, more CO₂ dissolves into the formation water [17,22]. Table 1 shows the summary of various chemical and physical trapping mechanism characteristics.

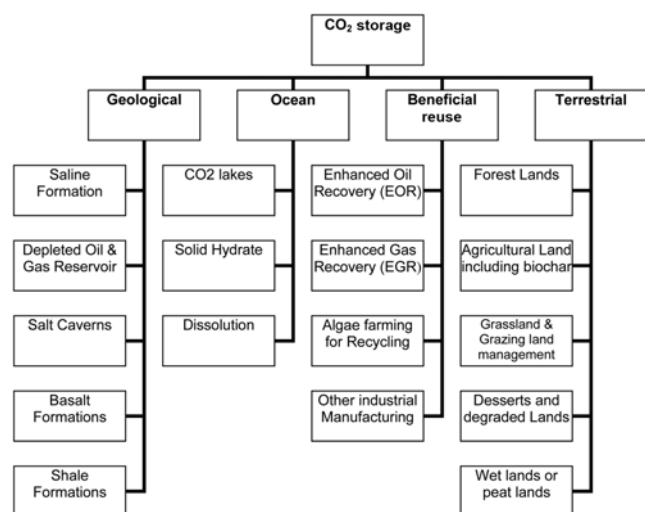
POSSIBLE STORAGE SITES

CO₂ can be stored in geological media by injection into oil and

[†]To whom correspondence should be addressed.
E-mail: chhan@snu.ac.kr

Table 1. Characteristics of trapping mechanism (Adapted from Bradshaw et al., 2007 [28])

Trapping Mechanism	Nature of trapping	Effective time frame	Areal size	Occurrence in basins
Structural and Stratigraphic	Buoyancy trapping within anticline, fold, fault block, pinch-out. CO ₂ remains as a fluid below physical trap (seal)	Immediate	10 to 100 s km	Dependent on basins tectonic evolution. Hundreds of small traps to single large traps per basin
Hydrodynamic	CO ₂ migrates through reservoir beneath seal, moving with or against the regional ground water flow system whilst other physical and chemical trapping mechanisms operate on the CO ₂	Immediate	Basin scale, e.g. 10,000 km	Along migration pathway of CO ₂ with or against the direction of the flow system that may move at rates of cm per year
Residual gas	CO ₂ fills interstices between pores of the grains of the rocks	Immediate to thousands of years	Basin scale, e.g. 1000 s km	Along migration pathway of CO ₂
Dissolution	CO ₂ migrates through reservoir beneath seal and eventually dissolves into formation fluid	100 to 1000 s of years if migrating more than 1000 s of years if gas cap in structural trap and longer if reservoir is thin and has low permeability	Basin scale, e.g. 10,000 s km	Along migration pathway of CO ₂ both up dip and down dip
Mineral precipitation	CO ₂ reacts with existing rock to form new stable minerals	10 to 1000 s of years	Basin scale, e.g. 10,000 s km	Along migration pathway of CO ₂
Coal adsorption	CO ₂ preferentially adsorbs onto coal surface	Immediate	10 to 100 s km	Limited to extent of thick coal seams in basins that are relatively shallow

**Fig. 1. Possible CO₂ storage options (Modified from Global CCS Institute Report, [30]).**

gas reservoirs, unminable coal seams and deep saline aquifers that are saturated with water whose salinity makes it unfit for human, animal, agricultural and industrial usage [3,29]. Fig. 1 shows the tree diagram of possible CO₂ storage sites, and Table 2 shows the key characteristics of some potential storage sites.

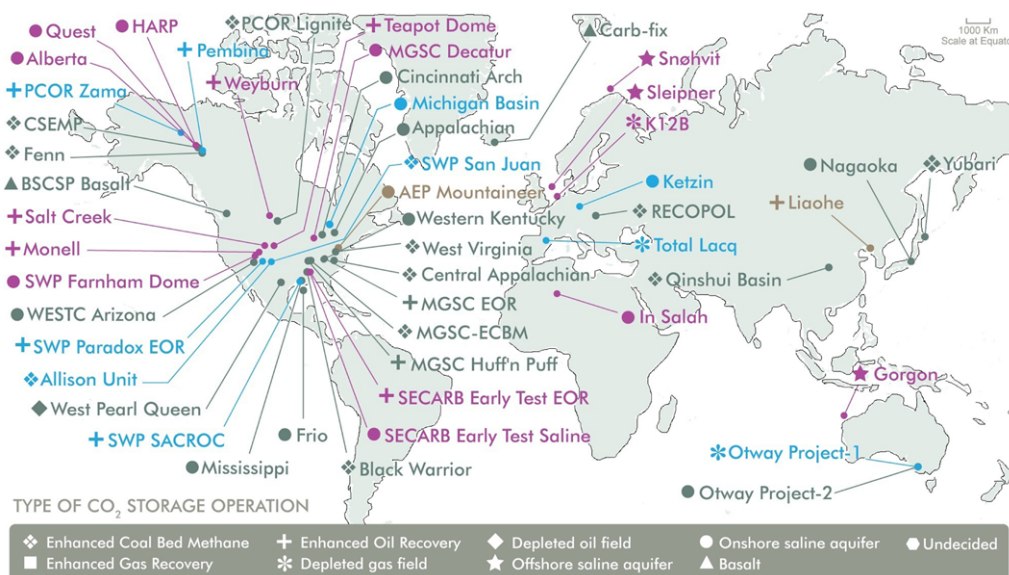
CURRENT SITUATION AND FUTURE OF STORAGE PROJECTS

Saline aquifers, which are found worldwide, present the largest capacity available for CO₂ storage [3]. By the end of 2008, approximately 20 Mton of CO₂ had been successfully injected into saline aquifers by existing operations. Currently, the highest injection rate and total storage volume for a single storage operation are approximately 1 M ton CO₂/year and 25 M ton, respectively [2]. Fig. 2 shows the active and planned CO₂ storage projects around the world for potential storage sites. The Global CCS Institute conducted a survey [30] in 2009 showing that within 219 active or planned CCS projects, 129 projects involve a form of storage as shown in Fig. 3. Of the projects active or planned for geological storage, 53% are for storage in a saline aquifer and 28% are for storage in depleted oil and gas fields. Of the projects considering beneficial reuse, 55% are for enhanced oil recovery (EOR) and 11% are for enhanced gas recovery (EGR)/enhanced coal bed methane (ECBM).

A study commissioned by the Global CCS Institute in 2010 [39] identified 80 largescale integrated projects at various stages of development around the world. The Global CCS institute categorized these storage projects as shown in Fig. 4. As of April 2010, public funding commitments were in the range of US\$ 26.6 billion to 36.1 billion [1]. Governments have announced their commitments to launch 19 to 43 large-scale integrated projects before 2020, details shown in

Table 2. Key characteristics of potential storage sites (Modified from 31)

	Advantages	Challenge	Project example	Data availability
Depleted oil & gas reservoir	- Global storage capacity of 140Gt and 40Gt for disused gas and oil fields respectively - Characteristics of reservoirs are well known - Existing infrastructure of wells and pipelines can be used - Proven containment over geologic time	- Concern over leaky wells or improperly abandoned wells- a safety threat - Today very few reservoirs depleted	CO2CRC Otway project [32]	Good
Saline aquifers	- Best potential CO₂ storage capacity (1000~10⁴ Gt) [3] - Stored CO₂ expected to be isolated from the near surface for thousands of years - Widespread presence over much of the world - Offshore aquifers eliminate most safety concerns	- Lack of characterization experience - Absence of financial incentive	- In Salah gas project, Algeria [33] - Sleipner project, Norway [34] - Gorgon project, Australia [35] - US DOE Regional partnership Program projects - Ketzin, Germany [36,37]	Good- in progress
Enhanced oil recovery (EOR)	- Additional oil, making it economically attractive - Injection of CO₂ commercially done today - Any undue risks not involved to humans or environment	- Could often be cheaper to obtain CO₂ from natural sources - Global storage capacity may be limited - For today's blow down reservoir operations need to store CO₂ under pressure	- Weyburn, Wason (Denver) - Wason (ODC) SACROC - Salt Creek	Excellent
Coal bed storage	- CH₄ by-product makes option economically attractive - Coal deposit present worldwide	Unminable coal seams are likely to be hundreds of meter deep, hence less permeable and limiting the capacity of CO₂ stored	- Qinshui Basin - Recopal - Alberta ECBM	Limited

**Fig. 2. Location of sites where activities relevant to CO₂ storage are planned or under way (as of April 2010) [38].**

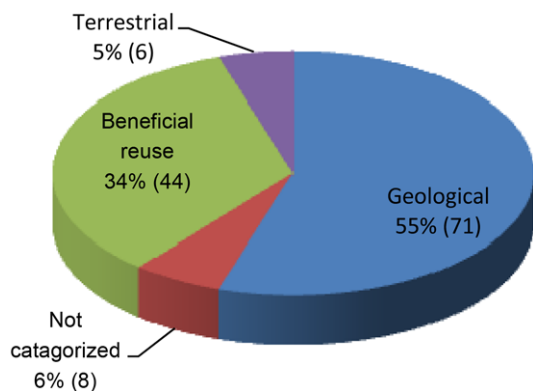


Fig. 3. Active or planned CCS projects by storage type [30].

Table 3. According to an IEA report [1], 39 projects involving storage are planned to be operational by 2015, which will be a major step in commercialization of CCS projects. This trend of rapid CCS implementation by 2015 is also prominent from the Global CCS Institute report [30], which is represented in graphical form in Fig. 5.

POTENTIAL SALINE AQUIFER STORAGE

Existing regional to basin scale capacity estimates are highly variable and in many instances contradictory. Bradshaw et al. [28] listed

Table 3. Funding and project announcement from governments and international organizations (as of April 2010) [1]

Country	Funding committed (US \$ Billion)	Number of projects committed by 2020
Australia	2 to 6	3 to 5
Canada	3.5	Upto 6
European commission	4 to 6	6 to 12
Japan	0.1	1 to 2
Norway	1	1 to 2
Korea	1	1 to 2
United kingdom	11 to 14.5	4
United states	4	5 to 10
Total	26.6 to 36.1	19 to 43

various CO₂ storage capacity estimates for the world and selected regions done under various studies. Koide et al. [40]; Hendricks et al. [41] presented an approach to estimate the CO₂ storage capacity considering only structural traps within a saline aquifer while Brunt et al. [42] presented capacity estimation taking in account the entire aquifer. Dooley et al. [43] compiled various storage capacities from different parts of world based on the data available and studies conducted previously. Fig. 6 represents the updated first-order theoretical storage capacity estimate. A technical study report

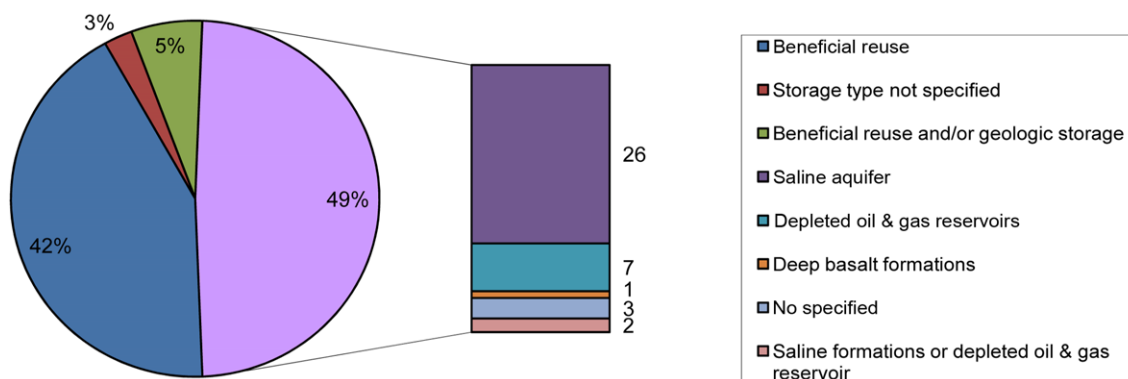


Fig. 4. Categorization of 80 active or planned storage projects (Modified from 39).

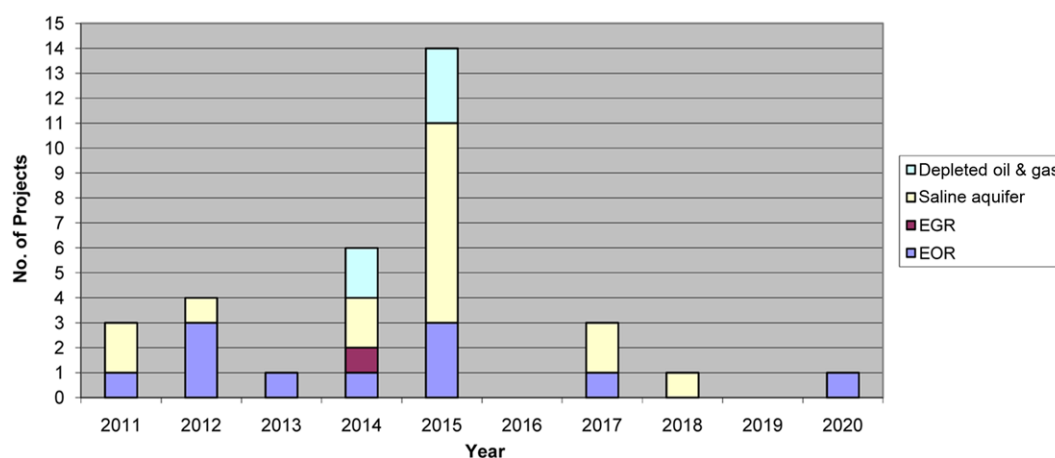


Fig. 5. Number of projects Vs storage type up to 2020 (Interoperated from Global CCS Institute report 2009, [30]).

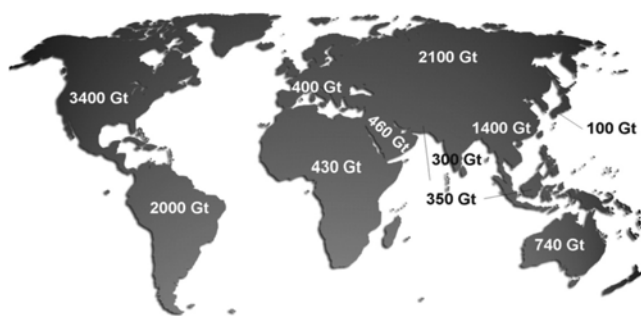


Fig. 6. Theoretical storage capacity estimate [44].

Table 4. Regional saline aquifer storage estimate

Country	Storage capacity (Gt)	Source
Brazil	2000	Ketzer et al. 2007 [45]
China	1435	Li 2007 [46]
Germany	20	May et al. 2005 [47]
Great Britain	14	Holloway et al. 2006 [48]
India	300	Sonde 2007 [49]
Japan	102	Suekane et al. 2008 [50]
Poland	5	Wojcicki et al. 2007 [51]
South Africa	20	Engelbrecht et al. 2004 [52]
US+ Canada	2150 (920-3380)	DOE 2007a [53]

by IEA [44] on saline aquifer storage presents regional saline aquifer capacity estimates as shown in Table 4.

CAPACITY ESTIMATION OF STORAGE SITES

The storage capacity of CO₂ is an estimate of the quantity of CO₂ that can be stored in a subsurface geologic formation. Because of uncertainties inherent in subsurface evaluation, exact quantification of geological properties is not possible, and therefore storage capacity is always at best an approximation of the amount of CO₂ that can be stored [44]. Because the CO₂ industry is not mature, there are few active CO₂ storage projects which can provide site specific information; hence low and high range capacity estimates are usually reported. Primarily, two methodologies are currently in use for estimation of storage capacity of CO₂, published by DOEa [53] and CSLF [54]. Detailed comparison of the two methodologies was performed by CSLF [55] and CO2CRC [56].

1. DOE Methodology [53,55-57]

The DOE defines static and dynamic methods for estimation of sub-surface storage volumes. The static methods are volumetric and compressibility based, while dynamic methods are applied after the injection is started. The volumetric method uses porosity (ϕ), area (A) and thickness (h) in a Monte Carlo simulation approach to estimate expected uncertainty in the efficiency of the storage capacity from a combination of trapping mechanisms [57]. Storage efficiency factor (E) provides a mean of estimating storage volume adjusting the uncertainty in capacity estimates.

Saline Aquifer	$G = A h_g \phi_i \rho E$	(1)
Oil & Gas Fields	$G = \rho A h_n \phi_e (1 - S_w) B E$	(2)

Coal Seams	$G = A h_g C \rho E$	(3)
------------	----------------------	-----

2. CSLF Methodology [54-56]

The use of CO₂ for EOR is a familiar technique but saline aquifers and deep unminable coal seams generally have limited data availability to estimate storage capacity volume. There are number of trapping mechanisms for geological storage of CO₂ in saline aquifers namely, structural and stratigraphic trapping, residual gas saturation trapping, dissolution, precipitation and hydrodynamic trapping. In reality, the various trapping mechanisms do not operate in isolation but in complex, interdependent and time-dependent ways. So, CSLF methodology differentiates capacity calculation for different trapping mechanisms.

Saline Aquifer	Structural & Stratigraphic trap	$V_t = A \phi h (1 - S_{wirr})$	(4)
		$V_e = C_c V_t$	(5)
	Solubility trap	$M_t = \iiint \phi (\rho_s X_s^{CO_2} - \rho_0 X_0^{CO_2}) dx dy dz$	(6)
		$M_t = A \phi h (\rho_s X_s^{CO_2} - \rho_0 X_0^{CO_2})$	(7)
		$M_e = C M_t$	(8)
	Residual trap	$V_t = \Delta V_{trap} \phi S_{CO_2}$	(9)
Gas Fields		$M_t = \rho_{CO_2} R_f (1 - F_{IG}) OGIP [(P_s Z_r T_r)/(P_s Z_s T_s)]$	(10)
Oil Fields		$M_t = \rho_{CO_2} [R_f^* OOIP/B_f - V_{iW} + V_{pw}]$	(11)
Oil & Gas reservoir based on geometry		$M_t = \rho_{CO_2} [R_{f,A} h \phi (1 - S_w) - V_{iW} + V_{pw}]$	(12)
		$M_e = C_m C_b C_h C_w C_a M_t = C_e M_t$	(13)
Coal Seams		$M_t = \rho_{CO_2} A h n_c G_c (1 - f_a - f_m)$	(14)
		$G_{cs} = V_L^* (P/P + P_L)$	(15)
		$M_e = R_f C' M_t$	(16)

3. Comparison of Methodologies

The methodologies proposed by DOE and CSLF can be compared to each other if the assumptions made are the same. In case of CO₂ storage in coal beds, both the DOE and CSLF recommend that the maximum depth to be considered is where coal permeability becomes less than 1 mD. For CO₂ sequestration in oil and gas reservoirs, the fundamental assumption is that the volume previously occupied by hydrocarbon is available for CO₂ storage, which is in agreement with both DOE and CSLF proposed approaches, assumptions and methodologies [55]. CSLF provides individual equations for each of the trapping mechanisms, but the coefficients for storage capacity efficiency have not yet been determined. If the CSLF methodology resolves the storage efficiency factors, then perhaps it will be the more viable option, but until then the use of the DOE proposed methodology, which already includes efficiency factors, is recommended [56].

4. Other Researches

$$G_{CO_2} = A * h * \phi * \rho_{CO_2} * 0.01 * 0.02 * 10^{-12} \quad (17)$$

Hendriks et al. [41] suggested Eq. (17) to estimate storage capacity of CO₂ in saline aquifers. This equation represents a simplified version of the DOE proposed methodology. Multiplication by 0.01 and 0.02 assumes that 1% of the aquifer is part of a structural trap and 2% sweep efficiency, respectively.

MONITORING TECHNIQUES

Reliable and cost-effective monitoring will be an important part of making geologic sequestration a safe, effective and acceptable method for CO₂ control. Monitoring will be required as part of the permitting process for underground injection and will be used for a number of purposes, such as tracking the location of the plume of injected CO₂, ensuring that injection and abandoned wells are not leaking, and verifying the quantity of CO₂ that has been injected underground. Monitoring is essential to making geological storage reliable and safe. By itself, monitoring cannot guarantee safety, but it determines safe storage of injected CO₂ [58].

The DOE defines atmospheric, near-surface and sub-surface monitoring techniques in the report "Monitoring, Verification, and Accounting of CO₂ Stored in Deep Geologic Formations," [58]; a brief summary of these techniques is presented here:

1. Atmospheric Monitoring Techniques

Monitoring Technique	Function
CO ₂ Detectors	Sensors for monitoring CO ₂ in air
Eddy Covariance	Atmospheric flux measurement technique to measure atmospheric CO ₂ concentrations at a height above the ground surface.
Advanced leak detection system	Carried by aircraft or terrestrial vehicles for quantification of CO ₂ fluxes from the soil
Laser Systems and LIDAR	Open-path device that uses a laser to shine a beam with a wavelength that absorbs CO ₂
Tracers (Isotopes)	Natural isotopic composition and/or compounds injected into the target formation along with the CO ₂ to determine the flow direction and early leak detection.

2. Near Surface Monitoring Techniques

Monitoring Technique	Function
Ecosystem stress monitoring	Satellite or airplane-based optical method after emission has occurred
Tracers	CO ₂ soluble compounds injected along with the CO ₂ into the target formation used to determine the hydrologic properties, flow direction and low-mass leak detection.
Groundwater monitoring	Sampling of water or vadose zone/soil (near surface) for basic chemical analysis for early detection prior to large emissions.
Thermal Hyperspectral Imaging	An aerial remote-sensing approach primarily for enhanced coal bed methane recovery and sequestration.
Synthetic Aperture Radar (SAR & InSAR)	A satellite-based technology in which radar waves are sent to the ground to detect surface deformation.

Color Infrared (CIR) Transparency Films	A vegetative stress technology deployed on satellites or aerially to indicate vegetative health, which can be an indicator of CO ₂ or brine leakage.
Tiltmeter	Measures small changes in elevation via mapping tilt. Mature oil field technology for monitoring stream or water injection, CO ₂ flooding and hydrofracturing.
Flux Accumulation Chamber	Quantifies the CO ₂ flux from the soil, but only from a small, predefined area
Induced Polarization	Geophysical imaging technology commonly used in conjunction with DC resistivity to distinguish metallic minerals and conductive aquifers from clay minerals in subsurface materials
Spontaneous (Self) Potential	Measurement of natural potential differences resulting from electrochemical reactions in the subsurface.
Soil and Vadose Zone Gas Monitoring	CO ₂ retained in soil gasses provides a longer residence time. Detection of elevated CO ₂ concentrations well above background levels provides indication of leak and migration from the target reservoir.
Shallow 2-D Seismic	High resolution images showing the presence of gas phase CO ₂ .

3. Subsurface Monitoring Techniques

Monitoring Technique	Function
Multi-component 3-D Surface Seismic Time-lapse Survey	This technology can provide high-quality information on distribution and migration of CO ₂
Vertical Seismic Profile (VSP)	This technology that can provide robust information on CO ₂ concentration and migration.
Magnetotelluric Sounding	Calculates changes in electromagnetic field resulting from variations in electrical properties of CO ₂ and formation fluids.
Electromagnetic Resistivity	Measures the electrical conductivity of the subsurface including soil, ground water, and rock.
Electromagnetic Induction Tomography (EMIT)	Wellbore measurement using a rock parameter, such as resistivity or temperature, to monitor fluid composition in wellbore and very useful for wellbore leakage
Annulus Pressure Monitoring	A mechanical integrity test on the annular volume of a well to detect leakage from the casing, packer or tubing.

Pulsed Neutron Capture	Capable of depicting oil saturation, lithology, porosity, oil, gas and water by implementing pulsed neutron techniques.
Electrical Resistance Tomography(ERT)	High resolution technique to monitor CO ₂ movement between wells
Sonic (Acoustic) Logging	Oil field technology used to characterize lithology, determine porosity, and travel time of the reservoir rock.
2-D Seismic Survey	Can be used to monitor “bright spots” of CO ₂ in the subsurface
Time-lapse Gravity	Use of gravity to monitor changes in density of fluid resulting from injection of CO ₂
Density Logging (RHOB Log)	Can estimate formation density and porosity at varying depths
Optical Logging	Optical imaging tools to provide detailed digital images of the well casing
Cement Bond Log (Ultrasonic Well Logging)	Implement sonic attenuation and travel time to determine whether casing is cemented or free. Allows for proactive remediation prior to leakage

Gamma Ray Logging	Use of natural gamma radiation to characterize the rock or sediment in a borehole
Microseismic (Passive) Survey	This technology can provide provides high-quality, high resolution subsurface characterization data and can provide effects of subsurface injection on geologic processes.
Crosswell Seismic Survey	Enables subsurface characterization between those wells
Aqueous Geochemistry	Chemical measurement of saline brine in storage reservoir.
Resistivity Log	Used to characterize the fluids and rock or sediment in a borehole.

Various monitoring is employed at different storage locations; Table 5 shows a summary of the monitoring techniques used at some projects.

SIMULATION TOOLS

Once the CO₂ is injected into a reservoir for storage, it behaves in a specific way depending on type of storage, storage conditions, fluid characteristics, injection rates and trapping mechanism involved. For safe, economical and efficient storage, it is important to understand and predict the behavior of CO₂ to be injected at a specific

Table 5. Summary of monitoring techniques at some projects

Project	Category	Monitoring techniques
Sleipner (2, 3, 59-69)	Saline aquifer	• Time-lapse seismic • Time-lapse gravity • Micro-seismic
Weyburn (44, 58, 70, 71)	EOR	• 4D, 9C surface seismic • 3D, 3C vertical seismic profile (VSP) • Cross-well seismic • Geochemical sampling analysis • Tracer injection monitoring • Conventional production data analysis • Passive seismic
Otway (3, 58, 71, 72)	Depleted gas field	• Lo Flo • Flux tower • Flask sampling • CO₂ sniffers • Headspace gas sampling • Surface soil gas • Hydrodynamic sampling • Groundwater chemistry • Downhole fluid sampling • VSP • 3D surface seismic • Borehole seismic • Microseismic • High resolution travel-time • Logging pressure/temperature

Table 6. Various simulator tools for CO₂ storage modeling (modified from 58)

Simulation tool name	Main sequestration application	Reference
GMI-SFIB, ABCUS	Modeling stresses applied to reservoirs during and after injection	DOE, US
Eclipse, GEM-GHG, NUFT	Model plume dispersion	Lindeberg et al. [73] Kumar et al. [74] Zhou et al. [75] Johnson et al. [76,77] Nghiem et al. [78]
PFLOTTRAN, STOMP	Model plume dispersion and CO ₂ interaction with reservoir fluids	White and Oostrom [79]
TOUGH-FLAC	Model plume dispersion and impact of stresses due to CO ₂ interactions	White et al. [80,81]
TOUGHREACT, VIP Reservoir	Model plume dispersion and CO ₂ trapping	Audigane et al. [59,82] Andre et al. [83] Xu et al. [84-86] Pruess and Garcia [87] Xu and Pruess [88]
NFFLOW-FRACGEN	CO ₂ flow through fractured networks	DOE, US
SIMED	Reservoir modeling of Coal Seam	Kreft et al. [89] Van der Meer et al. [90]
PHREEQC	Long term mineral trapping and porosity changes due to CO ₂ presence	Gunter et al. [91] Gaus et al. [92] Hellevang et al. [93]

site, guided by previous available data. To forecast the CO₂ behavior and expected pressure/temperature changes after injection, simulation tools are used. Table 6 shows the application of some simulation tools used in various studies. Simulations are used to predict temporal and spatial migration of the injected CO₂ plume, effect of geochemical reactions on CO₂ trapping and long-term porosity and

permeability, cap rock and wellbore integrity, the impact of thermal/compositional gradients in the reservoir, pathways of CO₂ out of the reservoir, the importance of secondary barriers, effects of unplanned hydraulic fracturing, the extent of upward migration of CO₂ along the outside of the well casing, impacts of cement dissolution and consequences of wellbore failure [51]. Effective monitoring can

Table 7. CO₂ storage cost estimates for different options

	On or offshore	Expected cost \$/t CO ₂ stored	Cost range US \$/t CO ₂ stored	Location	Reference
Saline aquifer	Onshore	0.5	0.5-5.1	Australia	Allinson et al. [94]
	Offshore	3.4	0.5-30.2	Australia	Allinson et al. [94]
	Onshore	2.5	1.9-6.2	US	Bock et al. [95]
	Onshore	2.8	1.9-6.2	Europe	Hendricks et al. [96]
	Offshore	7.7	4.7-12.0	Europe	Hendricks et al. [96]
Depleted oil and gas reservoir	Oil	1.3	0.5-4	US	Bock et al. [95]
	Gas	2.4	0.5-12.2	US	Bock et al. [95]
	Onshore	1.7	1.2-3.8	Europe	Hendricks et al. [96]
	Offshore	6.0	3.8-8.1	US	Bock et al. [95]
EOR		-14.8	-92~66.7	US	Bock et al. [95]
	Onshore		-10.5~10.5	Europe	Hendricks et al. [96]
	Offshore		-10.5~21.0		Hendricks et al. [96]
ECBM			-20.0~150.0		IEA [97]
	Onshore	-8.1	-26.4~11.1	US	Bock et al. [95]
			0~31.5	Europe	Hendricks et al. [96]
				Canada	Wong et al. [98]

verify whether the monitored results are the same as expected from the simulation model or not. This is mainly important in the early stages of a project to calibrate the model and alter the basis of simulation for long-term performance if required. Simulation tools have been used for research, pilot and commercial projects including the Weyburn, Frio, West Pearl Queen and ECBM West Virginia [51].

COST OF STORAGE

Storage cost of CO₂ depends on the type of storage option, location, depth and characteristics of the storage reservoir formation and the benefits and prices of any saleable products (in case of EOR, EGR, and ECBM). IPCC [3] estimated that the total cost for saline aquifer storage ranges from US\$ 0.5-8.0/ton CO₂ avoided including the monitoring cost of US\$ 0.1-0.3/ton CO₂. Table 7 shows the storage cost estimates for different locations.

Onshore storage cost depends on location, terrain and other geographic factors. The unit costs are usually higher for platforms or sub-sea facilities. Hendricks et al. [41] studied the injection cost for onshore and offshore aquifers at different depths as shown in Table 8. Table 9 presents the cost data for various active CO₂ storage projects.

CONCLUSION

Early CO₂ projects have been visible and their success will likely impact future CO₂ storage projects. By 2015, a number of more storage projects will be operational, which will be helpful in decreasing knowledge gaps. Beneficial reuse ways of storing CO₂ presents an economically feasible opportunity along with CO₂ atmospheric reduction. Presently, the number of operational EGR, ECBM projects and their data availability is limited, but they could be a potential CO₂ storage location in future. Previous CO₂ capacity estimates have some uncertainty and contradictory values; there is a need to close this gap before large-scale storage implementation since it could be a leakage threat. Monitoring techniques needs to be more mature

Table 8. Injection costs for onshore & offshore aquifer [41]

Depth (m)	Onshore aquifer US\$/t CO ₂ avoided	Offshore aquifer US\$/t CO ₂ avoided
1000	2.24	5.60
2000	3.36	9.08
3000	7.34	14.18

to have accurate data and avoid repeatability issues. Potential cost of geological storage is known, which depends on location, depth, reservoir characteristics and other associated factors.

ACKNOWLEDGEMENT

This research was supported by a grant from the Gas Plant R&D Center funded by the Ministry of Land, Transportation and Maritime Affairs (MLTM) of the Korean government.

NOMENCLATURE

A	: area that defines the region begin assessed for CO ₂ storage
B	: formation volume factor
B _f	: formation volume factor
C	: coefficient
C'	: completion factor
C _a	: aquifer strength coefficient
C _b	: buoyancy coefficient
C _c	: capacity coefficient
C _h	: heterogeneity coefficient
C _m	: mobility coefficient
C _w	: water saturation coefficient
E	: CO ₂ storage efficiency factor
F _{IG}	: fraction of gas injected
f _a	: ash weight fraction of coal
f _m	: moisture weight fraction of coal
G	: estimate of CO ₂ storage capacity
G _c	: coal gas content
Gcs	: coal gas content at saturation
h	: average thickness
h _g	: gross thickness of saline formation/ coal seam for which CO ₂ storage is assessed within the basin or region defined by area A
M _e	: effective storage capacity
M _t	: theoretical storage capacity
̄n _c	: bulk coal density
OGIP	: original gas in place
OOIP	: original oil in place
P	: total pressure
P _L	: langmuir pressure
P _r	: pressure at reservoir condition
P _s	: pressure at surface condition

Table 9. Cost of various commercial CO₂ storage projects [3,44,99]

Project	Cost	Injection start	Injection finish	Injection rate (ton/day)	Capacity (kTon)	Status
Sleipner	US\$ 96 M	1996		2700	20000	Operative
Weyburn	US\$ 1.1 Billion	2000		5000	20000	Operative
InSalah	US\$ 100 M	2004		3500	17000	Operative
Snohvit	US\$ 191 M	2008		2000	23000	Operative
Ketzin	US\$ 19 M	2008	2010	86	60	Operative
Otway	US\$ 38.10M**	2008	2010	150	100	Operative
Gorgon	US\$ 800.95 M*	2014		12300	129000	Under Construction

*Exchange rate: 1 USD=1.05 AUD

**Phase 1 injection: 2008-2010

R_f : recovery factor
 S_{CO_2} : trapped CO₂ saturation after flow reversal
 S_w : average water saturation within the total area A and net thickness h_n
 S_{wirr} : irreducible water saturation
 S_w : average water saturation
 T_r : temp at reservoir condition
 T_s : temp at surface condition
 V_e : effective storage volume
 V_L : langmuir volume
 V_t : theoretical storage volume
 V_{iw} : volume of injected water
 V_{pw} : volume of produced water
 V_{trap} : rock volume previously saturated with CO₂ that is invaded by water
 $X_0^{CO_2}$: initial CO₂ content in water formation
 $X_s^{CO_2}$: CO₂ content in water formation at saturation
 Z_r : gas compressibility at reservoir condition
 Z_s : gas compressibility at surface condition

Greek Letters

ϕ : porosity
 ϕ_t : average total porosity of entire saline formation over thickness h_g
 ϕ_e : average effective porosity over net thickness h_n
 ρ : density of CO₂ under (pressure, temperature) that represents storage conditions
 ρ_{CO_2} : density of CO₂
 ρ_{CO_2s} : CO₂ density at surface condition

Abbreviations and Acronyms

CCS : carbon capture and storage
 CIR : color Infrared
 CO2CRC : cooperative research centre for greenhouse gas technologies
 CSLF : carbon sequestration leadership forum
 DOE : department of Energy
 EOR : enhanced oil recovery
 ECBM : enhanced coal bed methane
 EGR : enhanced gas recovery
 EMIT : electromagnetic induction tomography
 ERT : electrical resistance tomography
 IEA : international energy agency
 IPCC : intergovernmental panel on climate change
 SAR : synthetic aperture radar
 VSP : vertical seismic profile

REFERENCES

1. IEA/CSLF Report to the Muskoka 2010 G8 Summit Carbon Capture and Storage Progress and Next Steps, OECD/IEA (2010).
2. K. Michael, A. Golab, V. Shulakova, J. Ennis-King, G. Allinson, S. Sharma and T. Aiken, *Int. J. Greenhouse Gas Control*, **4**, 659 (2010).
3. B. Metz, Davidson, H. C. de Coninck, M. Loos and L. A. Meyer (eds.), IPCC Special Report on Carbon Dioxide Capture and Storage (2005).
4. S. Reeves, A. Taillefert, L. Pekot and C. Clarkson, DOE Topical Report (February, 2003).
5. S. Reeves, D. Davis and A. Oudinot., DOE Topical Report, (March, 2004).
6. S. Reeves, *Proceedings of the 7th international conference on greenhouse gas control technologies (GHGT-7)*, Vancouver, Canada, v.II, 1399 (2004).
7. A. Saghaifi, *Int. J. Coal Geology*, **82**, 240 (2010).
8. W. D. Gunter, T. Gentzis, B. A. Rottenfusser and R. J. H. Richardson, *Energy Convers. Manage.*, **38**, 217 (1997).
9. W. D. Gunter, M. J. Mavor and J. R. Robinson, *Proceedings of the 7th international conference on greenhouse gas control technologies (GHGT-7)*, Vancouver, Canada, v.I, 413 (2004).
10. S. Bachu, *Int. J. Greenhouse Gas Control*, **1**, 374 (2007).
11. J. Bradshaw, G. Allinson, B. E. Bradshaw, V. Nguyen, A. J. Rigg, L. Spencer and P. Wilson, *Proceedings of the 6th international conference on greenhouse gas control technologies (GHGT-6)*, Kyoto, Japan, v.I, 633 (2002).
12. S. Holloway and D. Savage, *Energy Convers. Manage.*, **34**, 925 (1993).
13. L. G. H. Van der Meer, *Energy Convers. Manage.*, **34**, 959 (1993).
14. V. B. Meer, *Oil & Gas Sci. Technol.*, **3**, 527 (2005).
15. S. Bachu, *Energy Convers. Manage.*, **41**, 953 (2000).
16. S. Bachu, *Environ. Geol.*, **44**, 277 (2003).
17. J. Ennis-King and L. Paterson, *Proceedings of the fifth international conference on greenhouse gas control technologies*, Cairns, Australia, 290 (2000).
18. J. Ennis-King and L. Paterson, *SPE Asia Pacific Oil and Gas Conference and Exhibition*, Melbourne, Australia (2002).
19. S. Holloway and R. van der Straaten, *Energy Convers. Manage.*, **36**, 519 (1995).
20. P. J. Cook, A. J. Rigg and J. Bradshaw, *APPEA J.*, **40**, 654 (2000).
21. K. T. Biddle and C. C. Wielchowsky, *American Association of Petroleum Geologists*, **60**, 219 (1994).
22. S. Bachu, W. D. Gunter and E. H. Perkins, *Energy Convers. Manage.*, **35**, 269 (1994).
23. M. H. Holtz, SPE Gas Technology Symposium, Calgary, Alberta, Canada (2002).
24. M. A. Flett, R. M. Gurton and I. J. Taggart, *Proceedings of the 7th international conference on greenhouse gas control technologies*, v.I, Vancouver, Canada. 501 (2004).
25. H. Koide, Y. Tazaki, Y. Noguchi, S. Nakayama, M. Iijima, K. Ito and Y. Shindo, *Energy Convers. Manage.*, **33**, 619 (1992).
26. K. Pruess and J. Garcia, *Environ. Geol.*, **42**, 282 (2002).
27. W. D. Gunter, E. H. Perkins and T. J. McCann, *Energy Convers. Manage.*, **34**, 941 (1993).
28. J. Bradshaw, S. Bachu, D. Bonijoly, R. Burruss, S. Holloway, N. Peter Christensen and O. M. Mathiassen, *Int. J. Greenhouse Gas Control*, **1**, 62 (2007).
29. G. Aydin, I. Karakurt and K. Aydin, *Energy Policy*, **38**, 5072 (2010).
30. Global CCS Institute report, Strategic Analysis of the Global Status of Carbon Capture and Storage, Report 1: Status of Carbon Capture and Storage Projects Globally Final Report (2009).
31. Economic evaluation of CO₂ storage and sink enhancement options, EPRI, Palo Alto, CA, TVA, Muscle Shoals, AL, and U.S DOE, Washington DC (2002).
32. S. Sharma, P. Cook, T. Berly and M. Lees, *Energy Procedia*, **1**, 1965 (2009).

33. K. Michael, G. Allinson, A. Golab, S. Sharma and V. Shulakova, *Energy Procedia*, **1**, 1973 (2009).
34. <http://www.statoil.com>.
35. M. Flett, G. Beacher, J. Brantjes, A. Burt, C. Dauth, F. Koelmeyer, R. Lawrence, S. Leigh and J. McKenna, *Society of Petroleum Engineering*, SPE 116372 (2008).
36. F. Schilling, G. Borm, H. Würdemann, F. Möller and M. Kühn, *Energy Procedia*, **1**, 2029 (2009).
37. <http://www.co2sink.org/geninfo>.
38. http://www.co2crc.com.au/images/imagelibrary/gen_diag/world_projects_a_media.jpg.
39. Global CCS Institute Interim Report, Status of CCS Projects (2010).
40. H. Koide, Y. Tazaki, Y. Noguchi, M. Iijima, K. Ito and Y. Shinudo, *Energy Convers. Manage.*, **34**, 921 (1993).
41. C. Hendriks, W. Graus and F. van Bergen, *Global carbon dioxide storage potential and costs*, Report Ecofys & The Netherlands Institute of Applied Geoscience TNO, Ecofys Report EEP02002, 63 (2004).
42. R. G. Jr. Bruant, A. J. Guswa, M. A. Celia and C. A. Peters, *Environ. Sci. Technol.*, American Chemical Society, **7** (2002).
43. J. J. Dooley, R. T. Dahowski, C. L. Davidson, M. A. Wise, N. Gupta, S. H. Kim and E. L. Malone, Global Energy Technology Strategy program (GTSP), Battelle Memorial Institute, 37 (2006).
44. IEA Greenhouse Gas R&D Programme (IEA GHG), Aquifer Storage-Development Issues (2008).
45. J. M. Ketzer, J. A. Villwock, C. Caporale, L. H. da Rocha, G. Rockett, H. Braum and L. Giraffa, Sixth Annual Conference on Carbon capture and sequestration, Pittsburgh, Pennsylvania (2007).
46. X. Li, Joint Workshop on IGCC& Co-production and CO₂ capture & Storage, Beijing, Belfer Center for Science and International Affairs (2007).
47. F. May, C. Muller and C. Bernstone, *Int. J. Electr. Heat Gen.*, **85**, 32 (2005).
48. S. Holloway, C. J. Vincent, M. S. Benthams and K. L. Kirk, *Environ. Geosci.*, **13**, 71 (2006).
49. R. R. Sonde, Carbon Capture & Storage: Indian Perspective, CSLF Meeting, Paris (2007).
50. T. Suekane, T. Nobuso, S. Hirai and M. Kiyota, *Int. J. Greenhouse Gas Control*, **2**, 58 (2008).
51. A. Wojcicki, R. Tarkowski and B. Uliasz-Misiak, *Preliminary results of EU GeoCapacity project for Poland*, CO₂NET Annual Seminar, Lisbon (2007).
52. A. Engelbrecht, A. Golding, S. Hietkamp and B. Scholes, *The potential for sequestration of carbon dioxide in South Africa*, Pretoria, CSIR Manufacturing and Materials technology 54 (2004).
53. DOE 2007a, Carbon Sequestration Atlas of the United States and Canada, US Department of Energy/NETL, 88 (2007).
54. Estimation of CO₂ Storage Capacity in Geological Media - Phase II. Prepared by the Task Force on CO₂ Storage Capacity Estimation for the Technical Group (TG) of the Carbon Sequestration Leadership Forum (2007).
55. Comparison between Methodologies Recommended for Estimation of CO₂ Storage Capacity in Geological Media, in Geological Media by the CSLF Task Force on CO₂ Storage Capacity Estimation and the USDOE Capacity and Fairways Subgroup of the Regional Carbon Sequestration Partnerships Program - Phase III Report, CSLF-T-2008-04 (2008).
56. CO₂CRC, Storage Capacity Estimation, Site Selection and Characterization for CO₂ Storage Projects. Cooperative Research Centre for Greenhouse Gas Technologies, Canberra. CO₂CRC Report No. RPT08-1001, 52 (2008).
57. Methodology for Development of Geologic Storage Estimates for Carbon Dioxide Prepared for US Department of Energy National Energy Technology Laboratory Carbon Sequestration Program Prepared by Capacity and Fairways Subgroup Of the Geologic Working Group of the DOE Regional Carbon Sequestration Partnerships (2008).
58. DOE report, Monitoring, Verification, and Accounting of CO₂ Stored in Deep Geologic Formations DOE/NETL-311/081508 (2009).
59. P. Audigane, I. Gaus, I. Czernichowski-Lauriol, K. Pruess and T. Xu, *Am. J. Sci.*, **307**, 974 (2007).
60. R. Arts, O. Eiken, A. Chadwick, P. Zweigel, L. Van der Meer and B. Zinszner, *Energy*, **29**, 1383 (2004).
61. R. Arts, O. Eiken, P. Chadwick, A. Zweigel and B. van der Meer, G. Kirby, *The Geological Society*, **233**, 181 (2004).
62. R. Arts, A. Chadwick, O. Eiken, S. Thibeau and S. L. Nooner, Ten Years' Experience of Monitoring CO₂ Injection in the Utsira Sand at Sleipner, Offshore Norway: First Break, **26**, (2008).
63. R. Arts and P. Winthagen, Monitoring options for CO₂ storage. In: Thomas, D., Benson, S.M. (Eds.), Carbon Dioxide Capture for Storage in Deep Geologic Formations-Results from the CO₂ Capture Project, Elsevier, **2**, 1001 (2005).
64. A. Chadwick, R. Arts and O. Eiken, Proceedings of the 6th Petroleum Geology Conference (2005).
65. A. Ghaderi and M. Landro, Proceedings of the 7th Intl. Conf. on Greenhouse Gas Control Technologies, Vancouver, Canada (2004).
66. S. L. Nooner, O. Eiken, C. Hermanrud, G. S. Sasagawa, T. Stenvold and M. A. Zumberge, *Int. J. Greenhouse Gas Control*, **1**, 198 (2007).
67. T. A. Torp and J. Gale, *Energy*, **29**, 1361 (2004).
68. P. Zweigel, A. E. Lothe and E. B. Lindeberg, *AAPG Bull.*, **83**, 1346 (1999).
69. P. Zweigel, R. Arts, A. E. Lothe and E. B. Lindeberg, *The Geological Society*, **233**, 165 (2004).
70. *A technical basis for carbon dioxide storage*, Prepared by members of the CO₂ Capture Project (2009).
71. C. M. Gibson-Poole, S. Edwards, R. P. Langford and B. Vakarelov, Review of Geological Storage Opportunities for Carbon Capture and Storage (CCS) in Victoria. The Cooperative Research Centre for Greenhouse Gas Technologies (CO₂CRC), Australian School of Petroleum, The University of Adelaide, Adelaide. ICTPL Consultancy Report Number ICTPL-RPT06-0506 (2006).
72. <http://www.co2crc.com.au/otway/monitoring.html>.
73. E. Lindeberg and P. Bergmo, *Proceeding of the 6th international greenhouse gas control technologies*, v. I, Kyoto, 489 (2003).
74. M. Kumar, G. A. Noh, K. Pope, S. B. Sepehrmoori and L. W. Lakre, Reservoir Simulation of CO₂ Storage in Deep Saline Aquifers, SPE89343 (2004).
75. W. Zhou, M. Stenhouse, R. Arthur, S. Whittaker, D. Law, R. Chalaturnyk and W. Jazrawi, *Proceedings of the 7th international conference on greenhouse gas control technologies*, v.1, Vancouver, Canada (2004).
76. J. W. Johnson, J. J. Nitao, C. Steefel and K. G. Knaus, First Annual Conference on Carbon Sequestration, Washington (2001).
77. J. W. Johnson, J. J. Nitao and K. G. Knaus, *Geol. Soc.*, **233**, 107

- (2004).
78. L. Nghiem, P. Sammon, J. Grabenstetter and H. Ohkuma, *14th Symposium on improved oil recovery tulsa*, Oklahoma (2004).
79. M. D. White and M. Oostrom, STOMP Subsurface Transport Over Multiple Phase: User's Guide PNNL-15782 (UC-2010), Pacific Northwest National Laboratory, Richland, WA (2006).
80. S. P. White, G. Weir and W. Kissling, Proceedings of the 1st National Conference on Carbon Sequestration Washington DC (2001).
81. S. P. White, R. G. Allis, J. Moore, T. Chidsey, C. Morgan, W. Gwynn and M. Adams, *Chem. Geol.*, **217**, 387 (2005).
82. P. Audigane, I. Gaus, I. Czernichowski-Lauriol, K. Pruess and T. Xu, *Proceedings in TOUGH symposium san francisco*, USA (2006).
83. L. Andr, P. Audigane, M. Azaroual and A. Menjoz, *Energy Convers. Manage.*, **48**, 1782 (2007).
84. T. Xu, J. Apps and K. Pruess, *J. Geophysical Res.*, **108**, 2071 (2003).
85. T. Xu, J. Apps and K. Pruess, *Chem. Geol.*, **217**, 295 (2005).
86. T. Xu, J.A. Apps, K. Pruess and H. Yamamoto, *Chem. Geol.*, **242**, 319 (2007).
87. K. Pruess and J. Garcia, *Environ. Geol.*, **42**, 282 (2002).
88. T. Xu and K. Pruess, *American J. Sci.*, **301**, 16 (2001).
89. E. Kreft, C. Bernstone, R. Meyer, F. May, R. Arts, A. Obdam, R. Svensson, S. Eriksson, P. Durst, I. Gaus, B. van der Meer and C. Geel, *Int. J. Greenhouse Gas Control*, **1**, 69 (2007).
90. B. Van der Meer, E. Kreft, C. Geel and J. Hartman, *14th Europec biennial conference held in madrid*, Spain (2004).
91. W. D. Gunter, B. Wiwchar and E. H. Perkins, *Mineralogy and Petrology*, **59**, 121 (1997).
92. I. Gaus, C. Le Guern, J. Pearce, H. Pauwels, T. Shepherd, G. Hatziyannis and A. Metaxas, *Proceedings of the 7th international conference on greenhouse gas control technologies*, Vancouver, Canada, v.1, 561 (2004).
93. H. Hellevang, P. Aagaard, E. Oelkers and B. Kvamme, *Environ. Sci. Technol.*, **39**, 8281 (2005).
94. W. G. Allinson, D. N. Nguyen and J. Bradshaw, *APPEA J.*, 623 (2003).
95. B. Bock, R. Rhudy, H. Herzog, M. Klett, J. Davison, D. De la Torre Ugarte and D. Simbeck, Economic Evaluation of CO₂ Storage and Sink Options. DOE Research Report DE-FC26-00NT40937 (2003).
96. C. Hendriks, W. Graus and F. van Bergen, Global carbon dioxide storage potential and costs. Report Ecofys & The Netherland Institute of Applied Geoscience TNO, Ecofys Report EEP02002, 63 (2002).
97. IEA-GHG, Enhanced Coal Bed Methane Recovery with CO₂ Sequestration, IEA Greenhouse Gas R&D Programme, Report No. PH3/3, August, 139 (1998).
98. S. Wong, W. D. Gunter and J. Gale, *Proceedings of the 5th international conference on greenhouse gas control technologies (GHGT-5)*, Cairns, Australia (2000).
99. <http://www.zero2.no/projects/list-projects/>.